

Development of Liquid Scintillators Using New Fluorescent Compounds

Akira MUROI,

Shinya ARAKI, Noriyuki MIYAMOTO,

Yoshitaka KUNO, Akira SATO

Dept. of Physics, Osaka University

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Outline

- Introduction
 - *Motivation of development of liquid scintillator using new fluorescent compounds,*
 - *fluorescent compounds used in organic light emission device (OLED)*
- Measurements
 - *new fluorescent compounds(OLED compounds)*
 - *solubility,*
 - *emission and absorption spectra*
 - *liquid scintillators using the OLED compounds*
 - *Light yields,*
 - *decay constants*
- Summary and outlook

Motivation

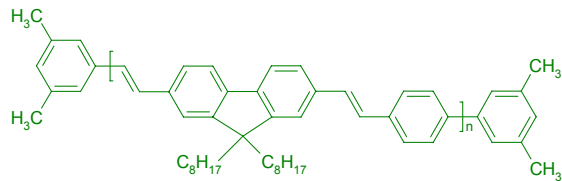
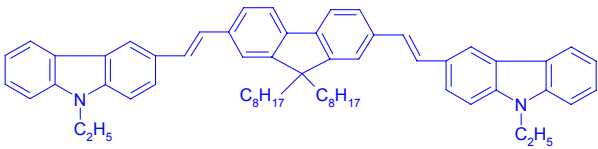
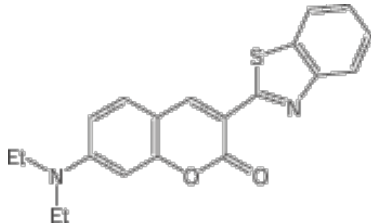


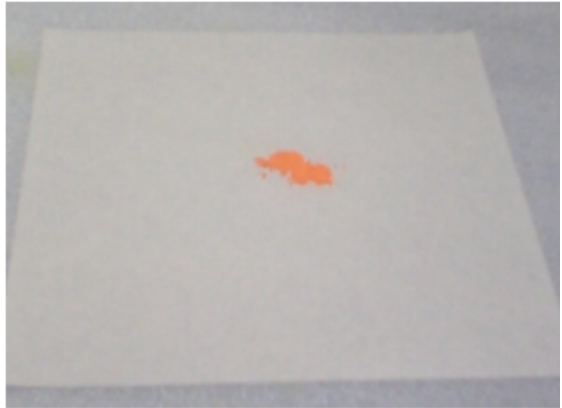
- *Recently, many new organic fluorescent compounds have been developed for organic light emitting devices (OLED), which are commercially used for flat-panel displays.*
- *Some of the fluorescent compounds have good properties to scintillators for HEP application, such as*
 - *high luminous efficiency,*
 - *blue or green light emitter, and*
 - *a short decay time constant (\sim nsec).*



... In This Study,

- *We investigate the properties of liquid scintillator cocktails using some of the compounds to study their potential for radiation detectors.*

Fluorescent compounds used in this study

ADS128GE	ADS086BE	Coumarin6
American Dye Source, Inc.	American Dye Source, Inc.	OHJEC Co.
		
Poly[(9,9-dioctylfluorenyl-2,7-diyl)-co-(1,4-divinylene-phenylene)]	1,4-Bis(9-ethyl-3-carbazovinylylene)-9,9-dihexylfluorene	3-(2-Benzothiazolyl)-7-(diethylamino)coumarin
		

Measurements

For each of OLED compounds,

- *solubility,*
- *emission and absorption spectra,*

For each of liquid scintillator cocktails using OLED compounds (OLED-LS),

- *light yields, and*
- *decay time constants*

were measured.

All the measurements were performed at a room temperature.

Solubility in Pseudocumene

Note; Pseudocumene (1,2,4-trimethylbenzene) is a typical solvent of liquid scintillator.

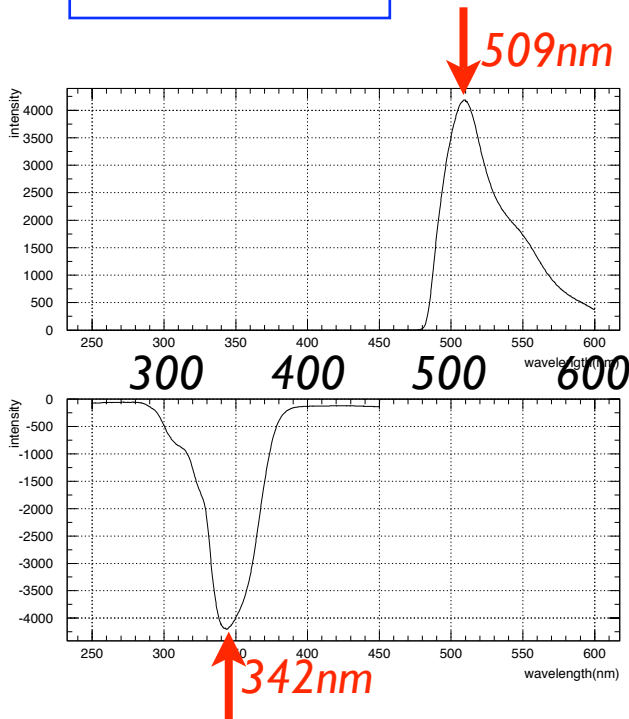
Compounds	solubility[g/l]@room temperature	
PPO	6.0	conventional fluorescent compounds for liquid scintillator
POPOP	0.07	
bis-MSB	0.75	
ADS086BE	3.8	OLED compounds
ADS128GE	<1.0	
Coumarin-6	<0.8	

These compounds have reasonable solubility, although it is smaller than PPO, which is typical primary solute.

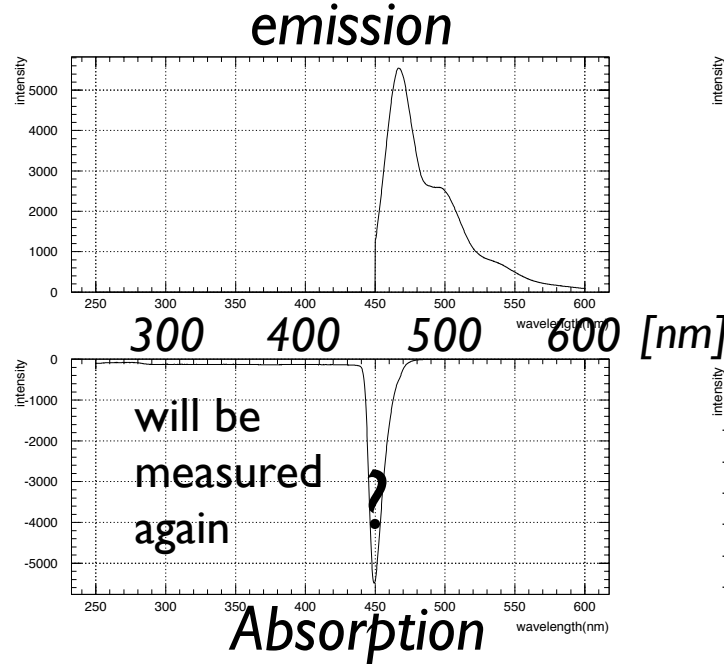
Emission and Absorption Spectra

in PseudoCumene

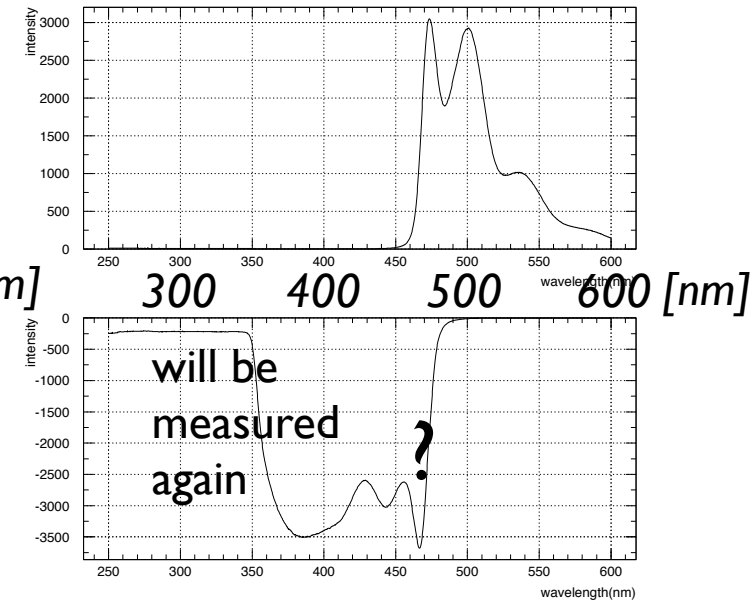
Coumarin-6



ADS086BE



ADS128GE



Upper Fig.: Emission spectra with the highest absorption wavelength

Lower Fig.: Absorption spectra with the highest emission wavelength

The absorption spectra of ADS086BE and ADS128GE will be measured again because of measurement mistakes (Raman scattering into a detector).

Wavelengths of Maximum Emission and Absorption

compounds	max.absorption	max.emission
PPO	352nm	375nm
POPOP	397nm	429nm
ADS086BE	(449nm)	467nm
ADS128GE	(383nm)	474nm
Coumarin6	342nm	509nm

- *These OLED compounds are seem to have absorption around the emission wavelength of PPO.*
- *It suggests that they can be excited by the emission of PPO, as a wavelength shifter.*

Primary and Secondary Solutes

- Basic process of energy transfer in organic scintillators
 - Charged particles interact with the **solvent** molecules
 - Energy transfer process occurs between the **solvent** and the **primary solute** molecules. [non-radiative]
 - Radiative transfer process between the **primary** and the **secondary solutes** occurs. Finally, the **secondary solutes** [wavelength shifter] emit photons with a longer wavelength than that of the **primary solutes**.

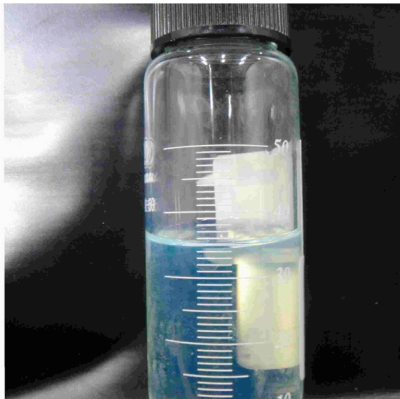




Generally, the concentration of the secondary solutes are often only 1% of that of the primary solute. In case of large-volume applications of liquid scintillators it is necessary to use a secondary solute.

Primary and Secondary Solutes

- *From the results of the solubility and the spectra measured, these OLED compounds can be used as a “wavelength shifter” in the scintillator solutions.*
- *Therefore, we prepared some liquid scintillators with*
 - *PPO as a primary solute, and*
 - *OLED compound as a secondary solute.*

Liquid Scintillator Cocktails with the OLED Compounds (OLED-LS)

<i>Solvent</i>	<i>Pseudocumene 40 ml</i>	<i>Pseudocumene 40 ml</i>	<i>Pseudocumene 40 ml</i>
<i>Primary solute</i>	<i>PPO 240 mg</i>	<i>PPO 240 mg</i>	<i>PPO 240 mg</i>
<i>Secondary solute</i>	<i>ADS086BE 150 mg</i>	<i>ADS128GE <30 mg</i>	<i>Coumarin6 <30 mg</i>
<i>Picture</i>			

BC-505 Liquid Scintillator as a Reference

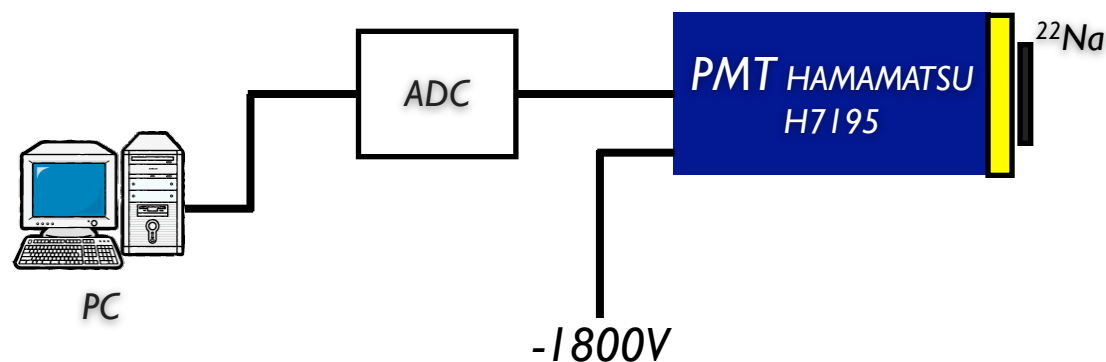


The BC-505 liquid scintillator from SAINT-GOBAIN CRYSTALS was used as a reference.

<i>Solvent</i>	<i>Pseudocumene (>90% of chemical composition)</i>
<i>Solutes</i>	<i>not known</i>
<i>Light output</i>	<i>80% of Anthracene</i>
<i>Wavelength of max. emission</i>	<i>425 nm</i>
<i>Decay constant</i>	<i>2.5 nsec</i>

Light Yield Measurements for OLED-LS

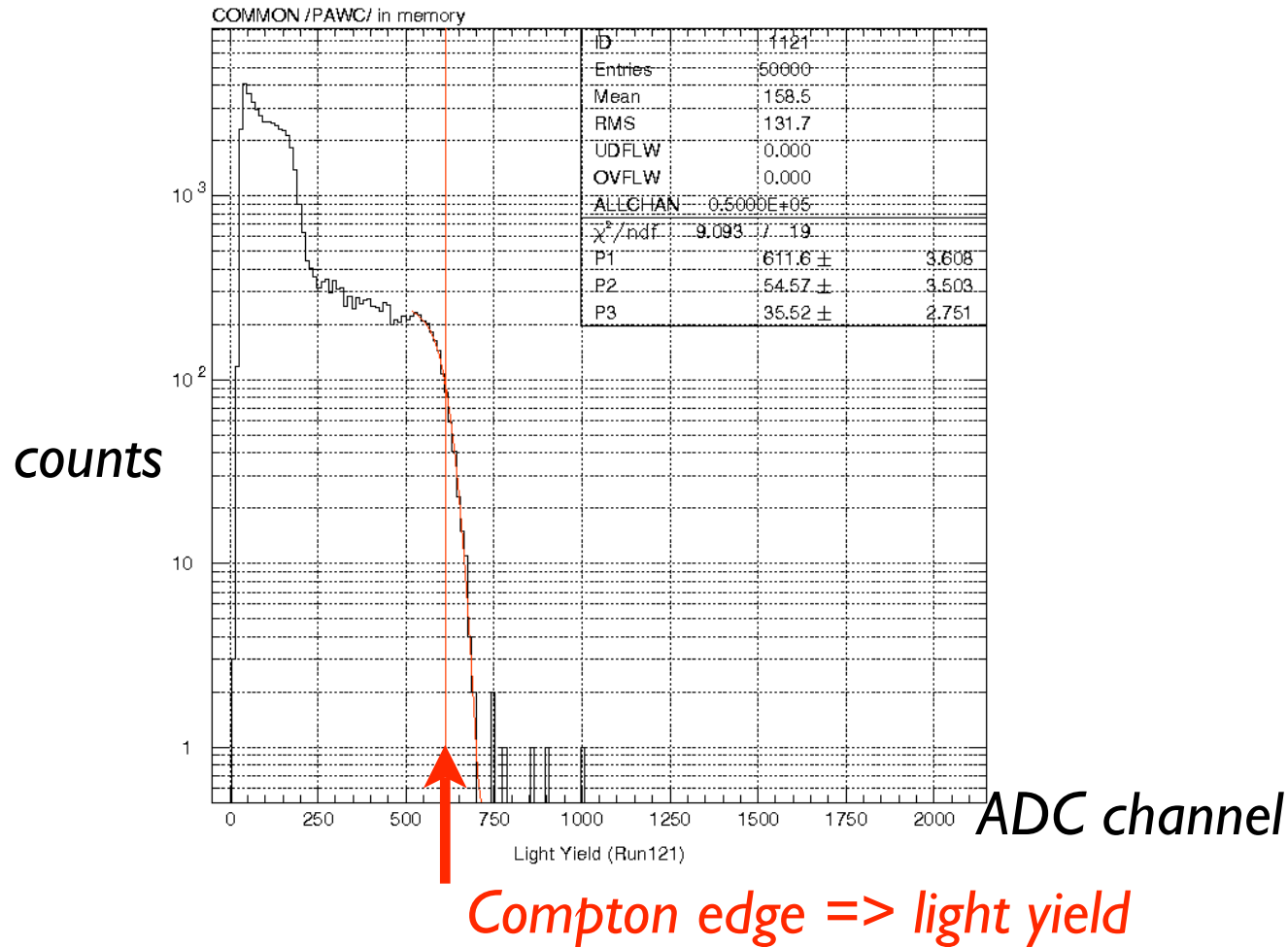
Experimental setup



quartz cell
10mm×10mm×50mm

The LS with the OLED compounds (OLED-LS) were excited by gamma-rays from a ^{22}Na source. A Hamamatsu H7195 PMT was used as a photon-detector. The LSs were filled in a quartz cell, and the PMT was placed directly on the cell. Pulse height spectra were recorded with CAMAC ADCs.

Typical ADC Spectra for ADS086BE

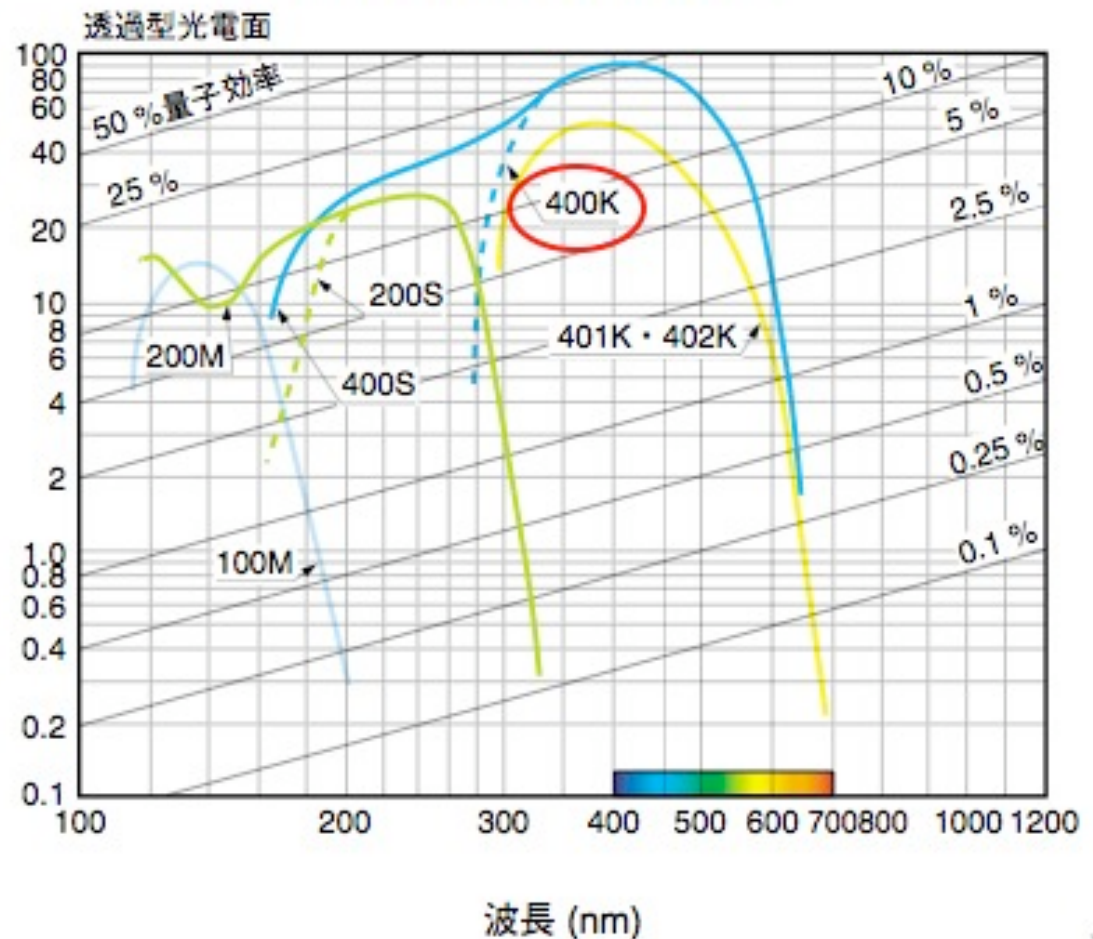


The ADC channels for the Compton edge were used as a measure of the light yield of the OLED-LS.

Quantum Efficiency Correction

Since different LSs have light emission of different wavelength, the light yields obtained from the measurements were corrected for quantum efficiency of the PMT over the range of fluorescence wavelength of the OLED-LS.

The comparison was made at wavelength of maximum emission for each.



Result of Light Yield Measurements

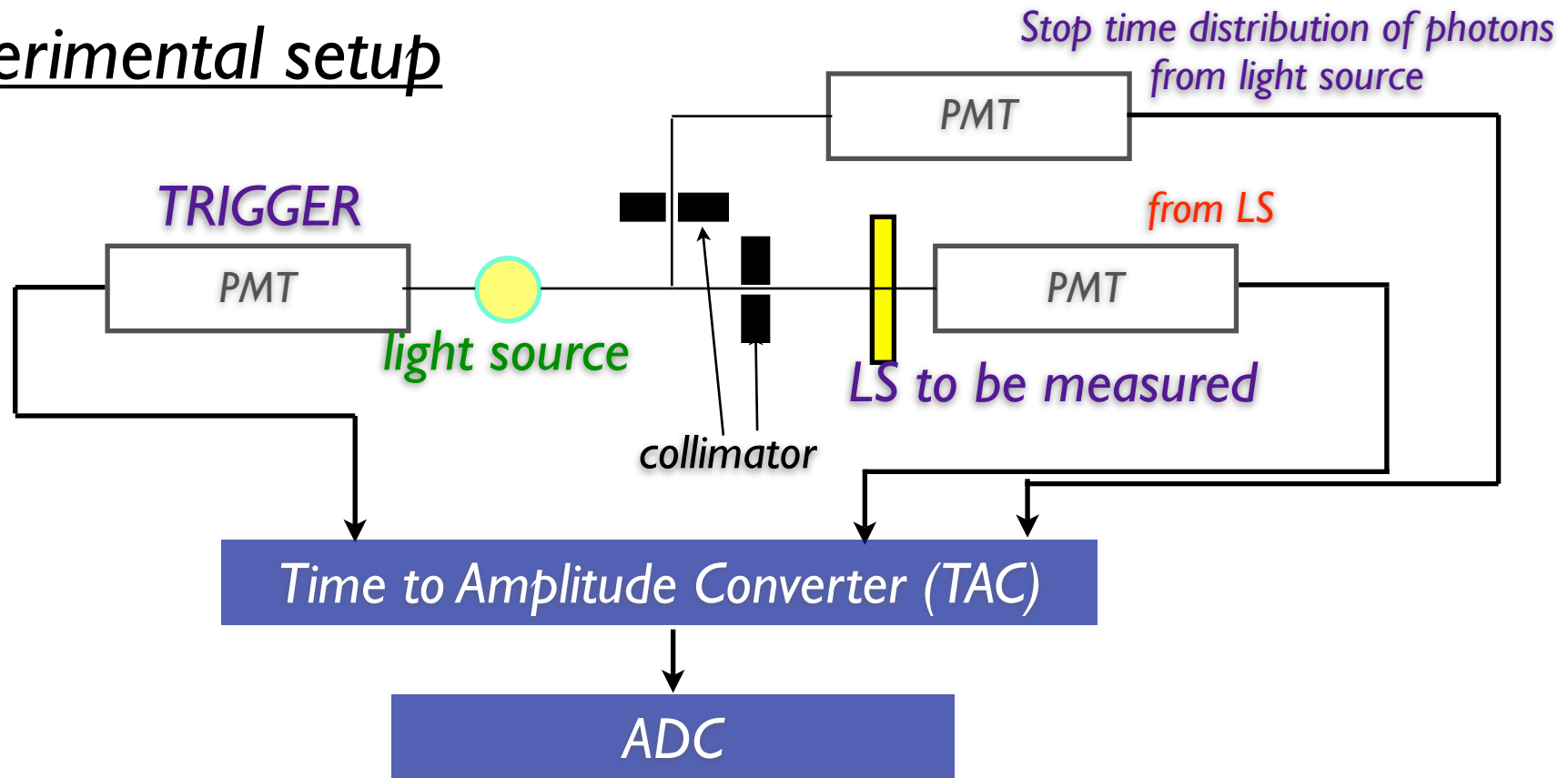
<i>LS</i>	<i>Compton edge [ch]</i>	<i>Relative QE correction</i>	<i>Relative Light yield [a.u.]</i>
<i>ADS086BE</i>	$284.7 \pm 8.5 \pm 1.6$	<i>1.3</i>	$110.7 \pm 3.0 \pm 0.6$ (at 467nm)
<i>BC505</i>	$332.8 \pm 8.9 \pm 1.6$	<i>1.0</i>	$100 \pm 2.7 \pm 0.5$ (at 429nm)

The relative light yields were obtained by multiplying the QE correction factor to the ADC channels of the compton edge.

The OLED-LS using ADS086BE has a 10% larger light yield than BC505!

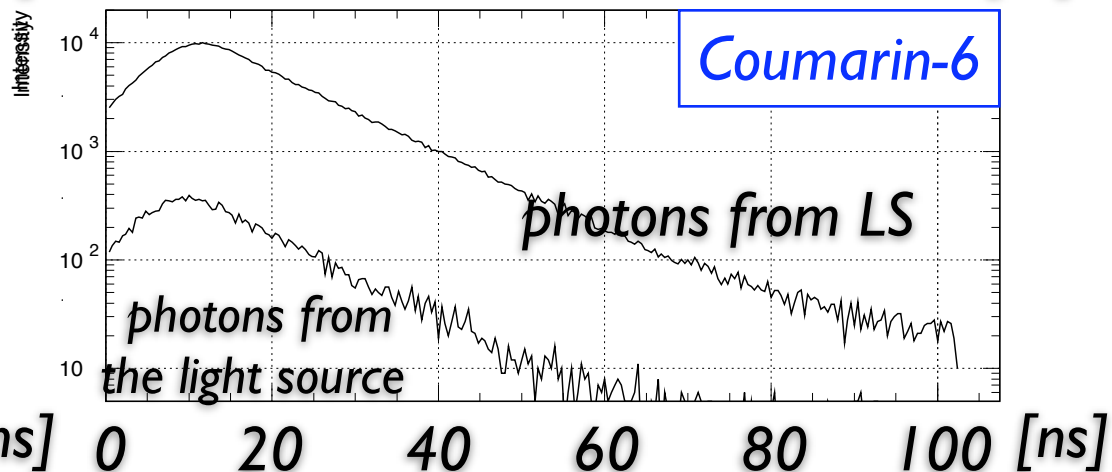
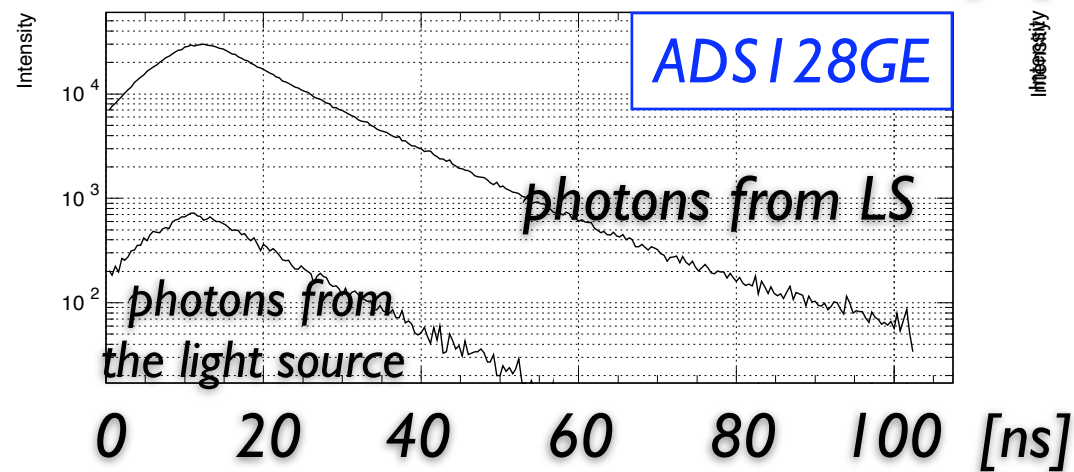
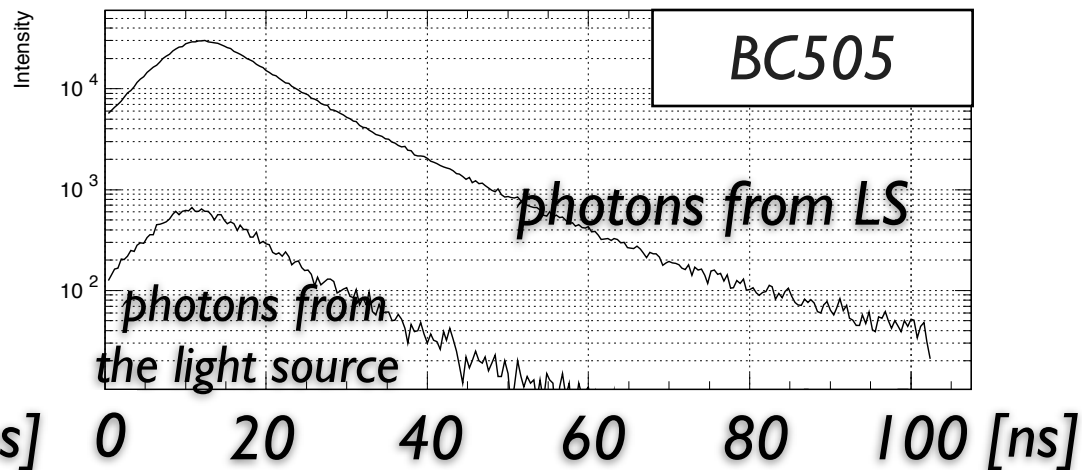
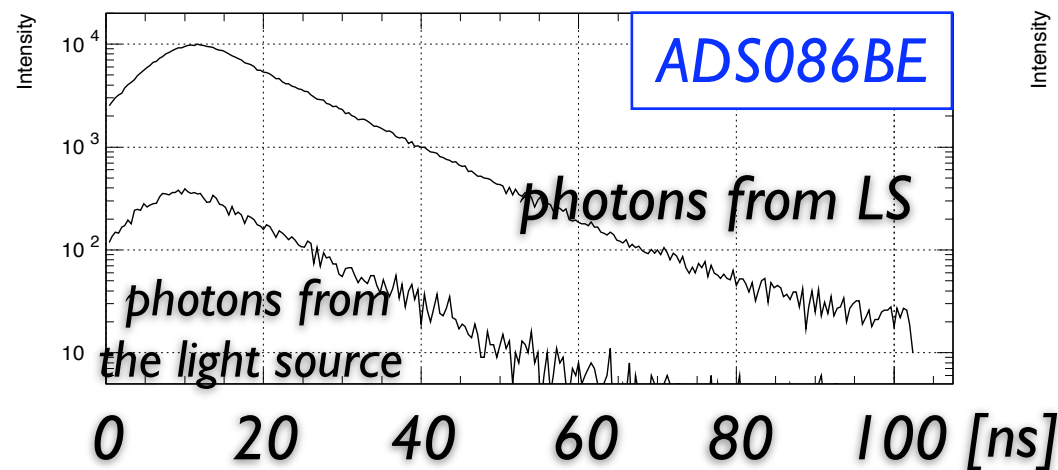
Decay Constant Measurement

Experimental setup



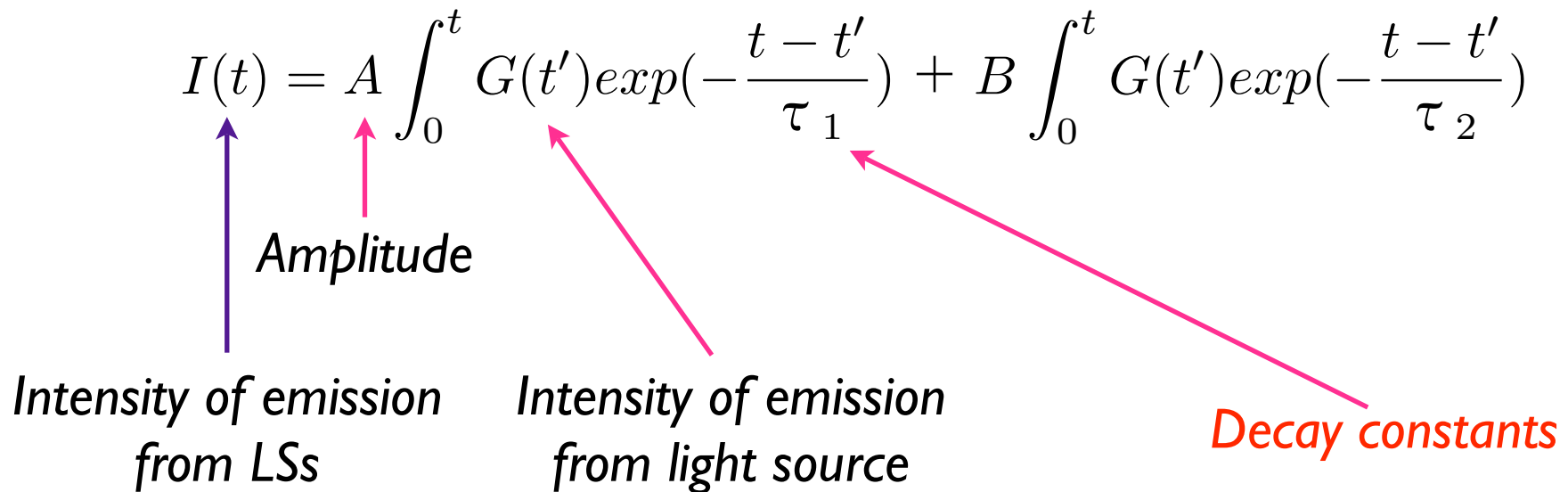
The emission time distributions were measured by the single photon counting technique, by using a Horiba NAES-550 setup for each OLED-LS cocktail.

Emission Time Distributions



Determination of the decay constants

To determine the decay constants, the emission time distributions were fitted to a multi-exponential function, and the light-source decay time was de-convoluted.

$$I(t) = A \int_0^t G(t') \exp\left(-\frac{t-t'}{\tau_1}\right) + B \int_0^t G(t') \exp\left(-\frac{t-t'}{\tau_2}\right)$$


The diagram illustrates the components of the multi-exponential function. A purple arrow points from the text "Intensity of emission from LSs" to the variable $I(t)$ in the equation. A pink arrow points from the text "Amplitude" to the coefficient A . Another pink arrow points from the text "Intensity of emission from light source" to the function $G(t')$. A red arrow points from the text "Decay constants" to the variables τ_1 and τ_2 in the exponential terms.

Result of Decay Constants Measurements

LS	BC505	ADS086BE	ADS128GE	Coumarin-6
Decay constant, main component [ns]	2.8 ± 0.1	$1.7 \pm 0.1^*$	$2.8 \pm 0.2^*$	$1.8 \pm 0.1^*$

* without Argon gas purge

The decay constant of some of the OLED-LS compounds (such as ADS086BE and Coumarin-6) have a faster decay constant than BC-505!

Summary of the LS measurements

LS	Wavelength of max. emission [nm]	Relative light yield with H7195 [a.u.]	Relative light yield after QE correction [a.u.]	Decay constant [ns]
BC-505	429	100 ±2.7±0.5	100 ±2.7±0.5 (80%Anthracene)	2.8±0.1
			100 ±2.7±0.5 (80% Anthracene)	
ADS086BE	467	83.6 ±2.5±0.4	108.3 ±3.2±0.6 (86%Anthracene)	1.7±0.1
			110.7 ±3.0±0.6 (88% Anthracene)	
ADS128GE	474	72.5 ±2.1±0.3	105.4 ±3.1±0.5 (84% Anthracene)	2.8±0.2
Coumarin-6	509	41.3 ±1.2±0.2	67.6 ±2.0±0.3 (54% Anthracene)	1.8±0.1

with
Ar purge

Future Plan and Outlook

- *The decay constants of OLED-LSs will be measured with Argon gas purge. They are expected to be better (longer) when Argon gas is not purged.*
- *Water-base scintillator using OLED fluorescent compounds will be investigated for large and stable detector application.*
- *Solid film or bulk scintillator using the fluorescent compounds would also be developed.*

Summary

- *Many kinds of new organic fluorescent compounds have been developed for commercial flat-panel displays, and they can be used for radiation detectors.*
- *To investigate their potential for making scintillators used in HEP application, their emission and absorption spectra and its solubility in pseudocumene have been measured. The compounds used in this study have shown they **can be used as a secondary solute in a liquid scintillator.***
- *For the liquid scintillator cocktails using the compounds (OLED-LS), the light yields and the decay constants have been determined.*
- *The light yield after the QE correction were **54-88 % Anthracene.** And, ADS086BE has shown a **10% larger light yield than BC-505.***
- *The decay constants of all of the OLED-LSs were **as fast as BC-505.***
- *The OLED fluorescent compounds would have high potential for new scintillator for the HEP use.*

Back up slides

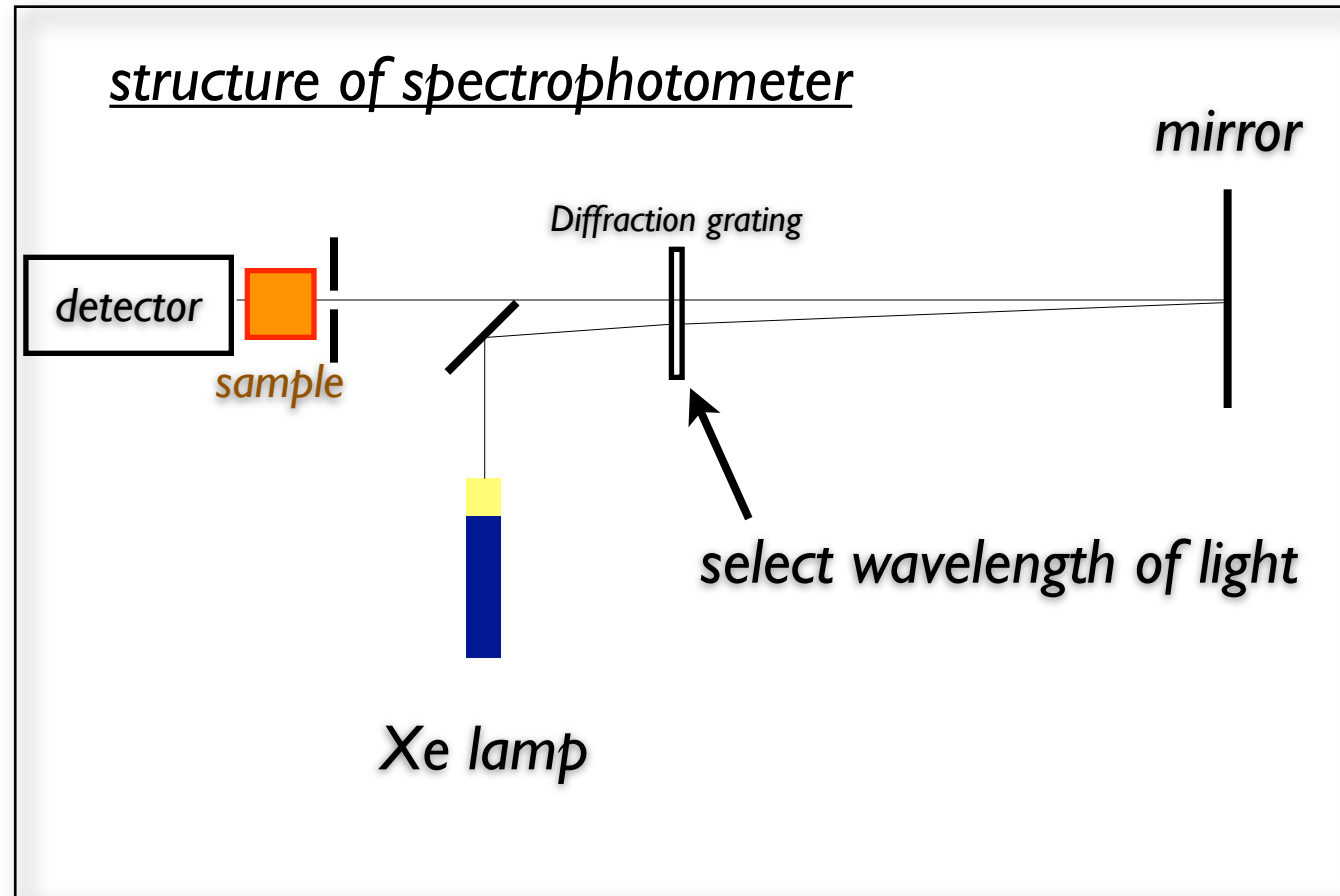
spectrophotometer for spectra measurement

HITACHI
spectrophotometer F-4500



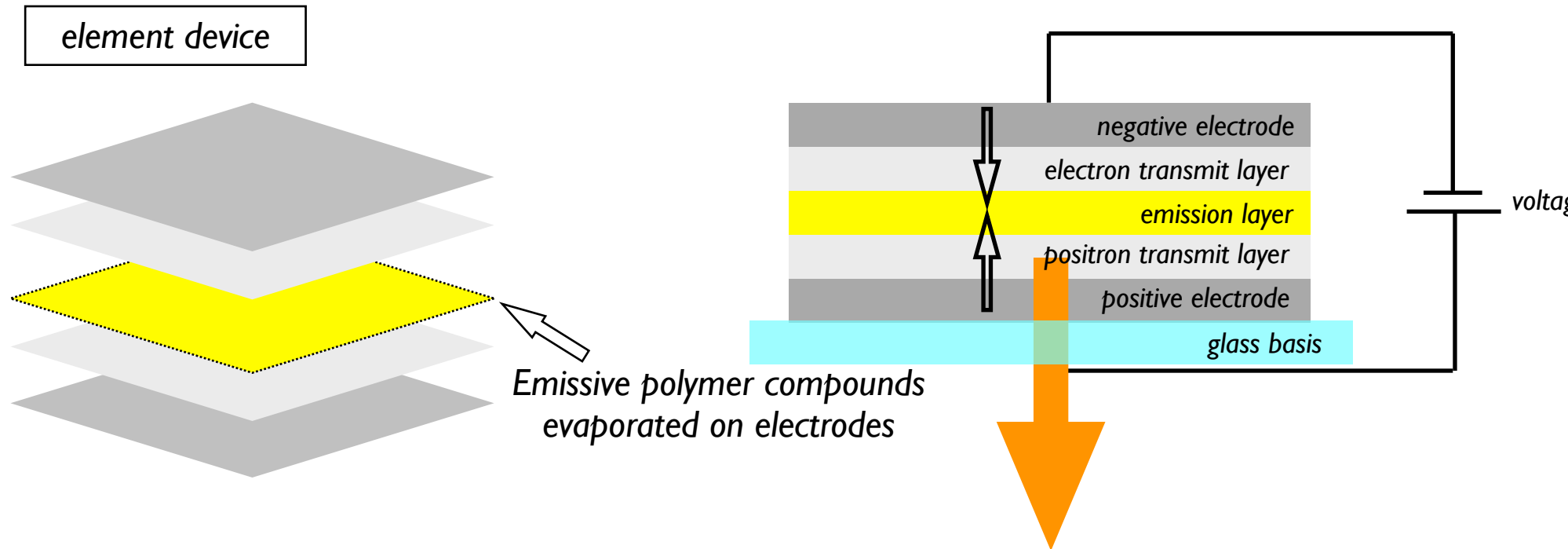
wavelength of light for absorption:
250nm~600nm

resolution:~1nm



HITACHI spectrophotometer F-4500 was used, which excites samples by light with wavelength of from 250nm to 600nm selected at a diffraction grating.

Element device of flat-panel display

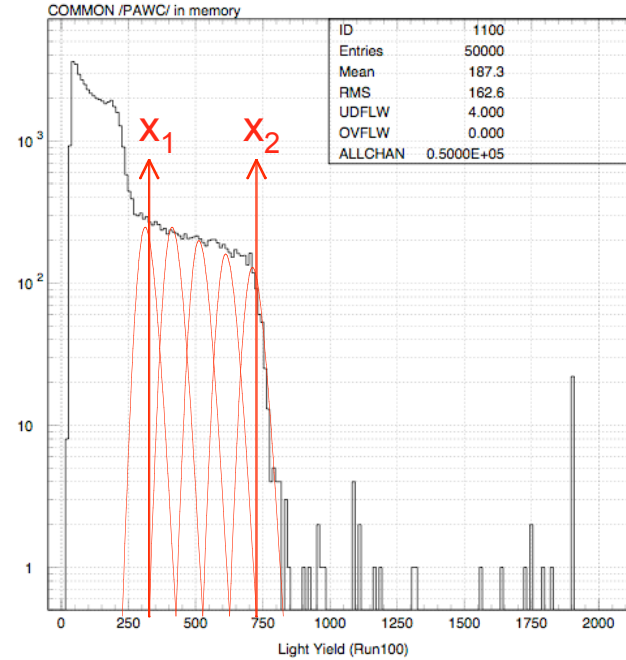


They are advertized as display which are operated with low voltage supplied, and which has luminosity and rapid response.



Determination of the Compton edge

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$$\text{Fitting func} = Ae^{-\frac{x}{b}} e\left(-\frac{1}{2} \left[\frac{(x-x_i)}{\sigma}\right]^2\right)$$

$X_1 \sim X_2$: